# organic compounds

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## (E)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

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Key indicators: single-crystal X-ray study; T = 113 K; mean  $\sigma(C-C) = 0.002$  Å; R factor = 0.027; wR factor = 0.076; data-to-parameter ratio = 13.4.

In the title compound  $C_{11}H_9Cl_2NO_2$ , the dihedral angle between the benzene and cyclopropane ring planes is 89.95 (13)°. The carbonyl-oxime grouping is almost coplanar with the benzene ring [dihedral angle =  $4.08 (6)^{\circ}$ ]. In the crystal, molecules are linked by  $C-H \cdots O$  interactions into [100] chains.

## **Related literature**

For further synthetic details, see: Liu et al. (2011b, 2012). For related structures, see: Liu & Liu (2011) Liu et al. (2011d). For the biological activity of related compounds, see: Liu et al. (2010, 2011a,c).



## **Experimental**

Crvstal data

$C_{11}H_9Cl_2NO_2$	$\gamma = 102.70 \ (3)^{\circ}$
$M_r = 258.09$	$V = 556.71 (19) \text{ Å}^3$
Triclinic, $P\overline{1}$	Z = 2
a = 6.4381 (13)  Å	Mo $K\alpha$ radiation
b = 7.6030 (15) Å	$\mu = 0.57 \text{ mm}^{-1}$
c = 11.956 (2) Å	T = 113  K
$\alpha = 94.90 \ (3)^{\circ}$	$0.24 \times 0.20 \times 0.10 \text{ mm}$
$\beta = 100.42 \ (3)^{\circ}$	

#### Data collection

Rigaku Saturn CCD diffractometer Absorption correction: multi-scan (CrystalClear; Rigaku/MSC, 2005) $T_{\min} = 0.876, T_{\max} = 0.946$ 

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.027$	145 parameters
$wR(F^2) = 0.076$	H-atom parameters constrained
S = 1.03	$\Delta \rho_{\rm max} = 0.24 \text{ e } \text{\AA}^{-3}$
1937 reflections	$\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$

3737 measured reflections

 $R_{\rm int} = 0.030$ 

1937 independent reflections

1563 reflections with  $I > 2\sigma(I)$ 

## Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C1-H1B\cdotsO1^{i}$	0.97	2.57	3.5008 (19)	161

Symmetry code: (i) x - 1, y, z.

Data collection: CrystalClear (Rigaku/MSC, 2005); cell refinement: CrystalClear; data reduction: CrystalClear; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6735).

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# supplementary materials

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## (E)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

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## Experimental

Dropwised the cyclopropanecarbonyl chloride to 2,4-dichlorobenzaldehyde oxime (7.50 mmol in 25 ml THF) and 7.5 mmol Et3N, then vigorously stirred at ambient temperature for overnight. The corresponding product precipitated immediately. Compound was dissolved in hot alcohol and the resulting solution was allowed to stand in air at room temperature to give colourless blocks of the title compound.

## Refinement

All the H atoms were positioned geometrically (C—H = 0.93–0.97 Å) and refined as riding with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(methyl C)$ .

## **Computing details**

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear* (Rigaku/MSC, 2005); data reduction: *CrystalClear* (Rigaku/MSC, 2005); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008.



## Figure 1

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level.

## (E)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

### Crystal data

 $\begin{array}{l} C_{11}H_9Cl_2NO_2\\ M_r = 258.09\\ Triclinic, P\overline{1}\\ a = 6.4381~(13)~\text{\AA}\\ b = 7.6030~(15)~\text{\AA}\\ c = 11.956~(2)~\text{\AA}\\ a = 94.90~(3)^\circ\\ \beta = 100.42~(3)^\circ\\ \gamma = 102.70~(3)^\circ\\ V = 556.71~(19)~\text{\AA}^3 \end{array}$ 

#### Data collection

Rigaku Saturn CCD	3737 measured reflections
diffractometer	1937 independent reflections
Radiation source: rotating anode	1563 reflections with $I > 2\sigma(I)$
Confocal monochromator	$R_{\rm int} = 0.030$
$\omega$ scans	$\theta_{\rm max} = 25.0^{\circ}, \ \theta_{\rm min} = 1.8^{\circ}$
Absorption correction: multi-scan	$h = -7 \rightarrow 7$
(CrystalClear; Rigaku/MSC, 2005)	$k = -9 \rightarrow 6$
$T_{\min} = 0.876, \ T_{\max} = 0.946$	$l = -14 \rightarrow 14$

## Refinement

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0407P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} = 0.001$
$\Delta \rho_{\rm max} = 0.24 \text{ e} \text{ Å}^{-3}$
$\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$

## Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Z = 2

F(000) = 264

 $\theta = 1.8 - 27.9^{\circ}$ 

 $\mu = 0.57 \text{ mm}^{-1}$ 

Block, coloress

 $0.24 \times 0.20 \times 0.10$  mm

T = 113 K

 $D_{\rm x} = 1.540 {\rm Mg} {\rm m}^{-3}$ 

Mo *K* $\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 1887 reflections

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
C11	0.32359 (6)	0.70035 (5)	0.47089 (3)	0.02317 (14)
Cl2	1.15092 (6)	0.98054 (6)	0.68823 (4)	0.02531 (14)
01	0.08882 (17)	0.30779 (16)	0.94813 (9)	0.0263 (3)
O2	0.00692 (16)	0.37138 (15)	0.76492 (9)	0.0186 (3)

N1	0.2306 (2)	0.46280 (18)	0.78101 (11)	0.0202 (3)
C1	-0.3868 (2)	0.2548 (2)	0.94150 (14)	0.0217 (4)
H1A	-0.2971	0.3293	1.0106	0.026*
H1B	-0.5348	0.2689	0.9234	0.026*
C2	-0.3504 (3)	0.0724 (2)	0.91336 (14)	0.0208 (4)
H2A	-0.4764	-0.0249	0.8783	0.025*
H2B	-0.2388	0.0354	0.9655	0.025*
C3	-0.2777 (2)	0.2198 (2)	0.84222 (14)	0.0198 (4)
H3	-0.3629	0.2113	0.7644	0.024*
C4	-0.0426 (2)	0.3004 (2)	0.86199 (13)	0.0177 (4)
C5	0.2637 (2)	0.5304 (2)	0.69054 (14)	0.0177 (4)
Н5	0.1511	0.5129	0.6268	0.021*
C6	0.4823 (2)	0.6365 (2)	0.68810 (14)	0.0164 (4)
C7	0.5265 (3)	0.7218 (2)	0.59348 (13)	0.0168 (4)
C8	0.7305 (2)	0.8271 (2)	0.59131 (13)	0.0188 (4)
H8	0.7566	0.8836	0.5275	0.023*
С9	0.8947 (2)	0.8453 (2)	0.68750 (14)	0.0187 (4)
C10	0.8595 (3)	0.7612 (2)	0.78232 (14)	0.0196 (4)
H10	0.9725	0.7736	0.8453	0.024*
C11	0.6539 (2)	0.6579 (2)	0.78256 (14)	0.0177 (4)
H11	0.6293	0.6016	0.8466	0.021*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0227 (2)	0.0314 (3)	0.0143 (2)	0.00420 (18)	0.00140 (17)	0.00786 (18)
Cl2	0.0192 (2)	0.0275 (2)	0.0273 (3)	-0.00120 (17)	0.00716 (17)	0.00581 (18)
01	0.0182 (6)	0.0425 (8)	0.0166 (7)	0.0021 (5)	0.0021 (5)	0.0121 (6)
O2	0.0123 (5)	0.0267 (6)	0.0165 (6)	0.0012 (5)	0.0038 (4)	0.0082 (5)
N1	0.0117 (6)	0.0267 (8)	0.0213 (8)	0.0005 (6)	0.0039 (5)	0.0072 (6)
C1	0.0179 (8)	0.0266 (9)	0.0221 (10)	0.0051 (7)	0.0073 (7)	0.0061 (7)
C2	0.0191 (8)	0.0226 (9)	0.0198 (9)	0.0004 (7)	0.0059 (7)	0.0054 (7)
C3	0.0151 (8)	0.0264 (9)	0.0160 (9)	0.0002 (7)	0.0027 (7)	0.0059 (7)
C4	0.0187 (8)	0.0201 (8)	0.0162 (9)	0.0045 (7)	0.0070 (7)	0.0059 (7)
C5	0.0178 (8)	0.0218 (9)	0.0145 (9)	0.0048 (7)	0.0041 (7)	0.0056 (7)
C6	0.0184 (8)	0.0158 (8)	0.0162 (9)	0.0042 (6)	0.0061 (7)	0.0031 (6)
C7	0.0185 (8)	0.0193 (8)	0.0134 (9)	0.0070 (6)	0.0019 (6)	0.0024 (6)
C8	0.0244 (9)	0.0189 (9)	0.0165 (10)	0.0057 (7)	0.0100 (7)	0.0071 (7)
C9	0.0181 (8)	0.0170 (8)	0.0223 (9)	0.0038 (7)	0.0082 (7)	0.0024 (7)
C10	0.0189 (8)	0.0226 (9)	0.0183 (9)	0.0077 (7)	0.0022 (7)	0.0040 (7)
C11	0.0194 (8)	0.0206 (9)	0.0166 (9)	0.0079 (7)	0.0065 (7)	0.0073 (7)

Geometric parameters (Å, °)

Cl1—C7	1.7477 (16)	C3—C4	1.470 (2)	
Cl2—C9	1.7394 (16)	С3—Н3	0.9800	
O1—C4	1.1976 (18)	C5—C6	1.467 (2)	
O2—C4	1.3800 (19)	С5—Н5	0.9300	
O2—N1	1.4253 (16)	С6—С7	1.394 (2)	
N1—C5	1.269 (2)	C6—C11	1.401 (2)	

C1—C2	1.478 (2)	С7—С8	1.385 (2)
C1—C3	1.517 (2)	C8—C9	1.389 (2)
C1—H1A	0.9700	С8—Н8	0.9300
C1—H1B	0.9700	C9—C10	1.378 (2)
C2—C3	1.508 (2)	C10—C11	1.383 (2)
C2—H2A	0.9700	C10—H10	0.9300
C2—H2B	0.9700	C11—H11	0.9300
C4 O2 N1	112 18 (11)	$O_2 C_4 C_3$	109 31 (12)
$C_{-}^{-}O_{-}^{-}N_{-}^{-}O_{-}^{-}$	109.05(12)	N1 - C5 - C6	109.31(12) 118.94(14)
$C_2 = C_1 = C_3$	60.45(11)	N1_C5_H5	120.5
$C_2 = C_1 = C_3$	1177	C6 C5 H5	120.5
$C_2 = C_1 = H_1 \Lambda$	117.7	$C_{0} = C_{0} = C_{0}$	120.3 117.60(14)
$C_2 = C_1 = H_1 R$	117.7	$C_{7} = C_{6} = C_{5}$	117.00(14) 121.71(14)
$C_2 = C_1 = H_1 B$	117.7	$C_{1} = C_{0} = C_{3}$	121.71(14) 120.60(15)
	11/./	$C^{\text{R}}_{\text{C}}$	120.09(13) 122.27(14)
$\Pi A - C I - \Pi B$	(114.0)	$C_{8} = C_{7} = C_{11}$	122.37(14) 116.74(12)
C1 = C2 = C3	01.08 (11)	$C_{0} = C_{1} = C_{1}$	110.74(13) 120.80(12)
C1 - C2 - H2A	1177	$C_0 - C_1 - C_1$	120.89 (12)
$C_3 = C_2 = H_2 A$	1177	$C_{1} = C_{2} = C_{2}$	117.79 (13)
$C_1 = C_2 = H_2 B$	1177	$C = C = H \delta$	121.1
$C_3 - C_2 - \Pi_2 B$	11/./	$C_{9}$	121.1
$H_2A = C_2 = H_2B$	114.8	C10 - C9 - C8	121.89 (15)
C4 - C3 - C2	116.62 (14)	C10 - C9 - C12	119.36 (12)
C4 - C3 - C1	115.92 (13)	$C_8 = C_9 = C_{12}$	118.75 (13)
$C_2 = C_3 = C_1$	58.47 (11)		119.14 (14)
C4 - C3 - H3	117.5	C9—C10—H10	120.4
C2—C3—H3	117.5	C11—C10—H10	120.4
CI = C3 = H3	117.5	C10-C11-C6	121.20 (15)
01	123.97 (14)	CIO—CII—HII	119.4
O1—C4—C3	126.72 (15)	С6—С11—Н11	119.4
C4—O2—N1—C5	-177.60 (13)	С5—С6—С7—С8	177.96 (16)
C1—C2—C3—C4	105.44 (16)	C11—C6—C7—Cl1	179.11 (13)
C2-C1-C3-C4	-106.64 (16)	C5—C6—C7—Cl1	-2.0 (2)
N1-02-C4-01	-3.1 (2)	C6—C7—C8—C9	0.4 (3)
N1-02-C4-C3	176.04 (13)	Cl1—C7—C8—C9	-179.65 (12)
C2—C3—C4—O1	-25.6 (3)	C7—C8—C9—C10	0.6 (3)
C1-C3-C4-O1	40.4 (2)	C7—C8—C9—Cl2	-178.69 (13)
C2—C3—C4—O2	155.33 (15)	C8—C9—C10—C11	-1.1 (3)
C1—C3—C4—O2	-138.66 (14)	Cl2—C9—C10—C11	178.24 (13)
O2—N1—C5—C6	178.12 (13)	C9—C10—C11—C6	0.5 (3)
N1—C5—C6—C7	-176.72 (16)	C7C6C11C10	0.5 (2)
N1-C5-C6-C11	2.2 (3)	C5-C6-C11-C10	-178.44 (16)
С11—С6—С7—С8	-1.0 (3)		
Hydrogen-bond geometry $(\hat{\lambda}^{0})$			

## *Hydrogen-bond geometry (Å, °)*

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$

D—H···A

# supplementary materials

C1—H1B····O1 <sup>i</sup>	0.97	2.57	3.5008 (19)	161	

Symmetry code: (i) x-1, y, z.